

## AP WORKSHEET 00D: Inorganic Nomenclature II

Add either a name or a formula to complete each table. (100)

1. Potassium dichromate	
2. Lithium sulfide	
3. Potassium bromide	
4. Cesium iodide	
5. Calcium phosphide	
6. Sodium fluoride	
7. Strontium oxide	
8. Beryllium sulfide	
9. Magnesium bromide	
10. Lithium oxide	
11. Strontium chloride	
12. Barium bromide	
13. Magnesium sulfide	
14. Magnesium iodide	
15. Hydrogen fluoride (Hydrogen monofluoride)	
16. Barium phosphide	
17. Sodium hydrogen phosphate	
18. Potassium chloride	
19. Lithium nitride	
20. Calcium sulfide	
21. Rubidium oxide	
22. Strontium nitride	
23. Cesium phosphide	
24. Magnesium carbonate	
25. Beryllium sulfate	

26. Dinitrogen Tetraoxide	
27. Carbon dioxide	
28. Mercury(I) chloride	
29. Hydroiodic acid	
30. Iodic acid	
31. Perbromic acid	
32. Hypobromous acid	
33. Phosphorus pentachloride	
34. Iodine monochloride	
35. Antimony(III) fluoride	
36. Bromine monofluoride	
37. Bromine dioxide	
38. Dinitrogen pentoxide	
39. Carbon monosulfide	
40. Tellurium dioxide	
41. Phosphorus tribromide	
42. Carbon tetraiodide	
43. Vanadium(V) chromate	
44. Zinc carbonate	
45. Silver hydroxide	
46. Vanadium(III) chromate	
47. Mercury(II) iodide	
48. Uranium(V) nitrate	
49. Nickel (III) nitride	
50. Sulfuric acid	

51. $\text{ScCl}_3$	
52. $\text{HCl}$	
53. $\text{PtO}_2$	
54. $\text{Sb}(\text{ClO}_3)_5$	
55. $\text{GeS}_2$	
56. $\text{ZnO}$	
57. $\text{VSO}_4$	
58. $\text{CuCl}_2$	
59. $\text{TiO}_2$	
60. $\text{NiN}$	
61. $\text{Ni}_3(\text{PO}_4)_2$	
62. $\text{CoF}_3$	
63. $\text{Au}_2\text{O}_3$	
64. $\text{Zn}_3\text{P}_2$	
65. $\text{Cr}(\text{NO}_3)_6$	
66. $\text{NaIO}_2$	
67. $\text{NaIO}_3$	
68. $\text{NaI}$	
69. $\text{H}_2\text{SO}_3$	
70. $\text{H}_2\text{CO}_3$	
71. $\text{AlN}$	
72. $\text{AlH}_3$	
73. $\text{Li}_3\text{AsO}_4$	
74. $\text{NaCN}$	
75. $\text{Na}_2\text{O}_2$	

76. $\text{Li}_3\text{PO}_3$	
77. $\text{KHCO}_3$	
78. $\text{HF}$	
79. $\text{AuI}_2$	
80. $\text{KMnO}_4$	
81. $\text{Na}_2\text{Cr}_2\text{O}_7$	
82. $\text{Ag}_2\text{CrO}_4$	
83. $\text{AgCl}$	
84. $\text{NaCH}_3\text{COO}$	
85. $\text{RaF}_2$	
86. $\text{KSCN}$	
87. $\text{FeS}$	
88. $\text{Fe}_2(\text{SO}_3)_3$	
89. $\text{FeSO}_4$	
90. $\text{MgS}$	
91. $\text{Na}_2\text{S}_2\text{O}_3$	
92. $\text{RbCl}$	
93. $\text{Cu}(\text{OH})_2$	
94. $\text{Mg}_3\text{N}_2$	
95. $\text{Cu}_3\text{N}$	
96. $\text{LiH}$	
97. $\text{K}_2\text{O}$	
98. $\text{K}_2\text{O}_2$	
99. $\text{Li}_3\text{N}$	
100. $\text{DsCl}_3$	