

## AP WORKSHEET 00D: ANSWERS

	Formula	Name
1	HClO <sub>2</sub>	Chlorous acid
2	NaBr	Sodium bromide
3	K <sub>2</sub> CO <sub>3</sub>	Potassium carbonate
4	Cu(HCO <sub>3</sub> ) <sub>2</sub>	Copper(II) hydrogencarbonate
5	Al <sub>2</sub> (SO <sub>3</sub> ) <sub>3</sub>	Aluminum sulfite
6	SrH <sub>2</sub>	Strontium hydride
7	BI <sub>3</sub>	Boron triiodide
8	N <sub>2</sub> O <sub>4</sub>	Dinitrogen tetroxide
9	SeF <sub>6</sub>	Selenium hexafluoride
10	P <sub>2</sub> O <sub>5</sub>	Diphosphorous pentoxide
11	FeCl <sub>2</sub>	Iron(II) chloride
12	OF	Oxygen monofluoride
13	Cu <sub>2</sub> SO <sub>4</sub>	Copper(I) sulfate
14	ClF <sub>5</sub>	Chlorine pentafluoride
15	MgSO <sub>4</sub>	Magnesium sulfate
16	MnO	Manganese(II) oxide
17	FeO	Iron(II) oxide
18	Mg(OH) <sub>2</sub>	Magnesium hydroxide
19	CrO <sub>3</sub>	Chromium(VI) oxide
20	Rb <sub>2</sub> O	Rubidium oxide